

# Carbohydrates

**Monosaccharides:**  
monomers and building blocks (glucose, fructose)

**Polysaccharides:** complex chain of monomers linked together (cellulose, starch, glycogen)

# Monosaccharides

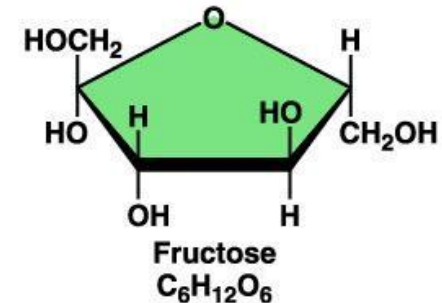
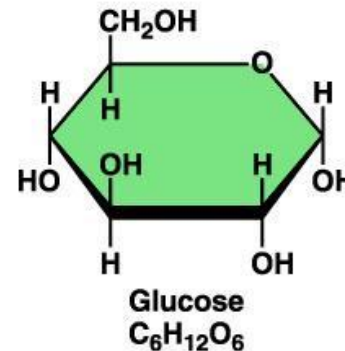
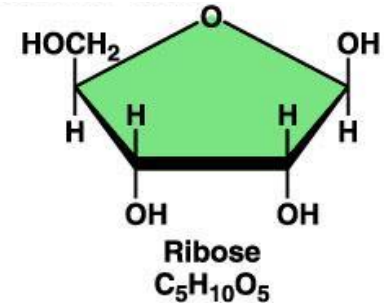
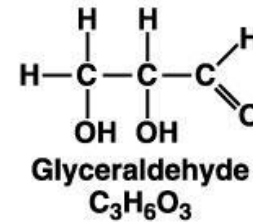
simple sugars containing 3 - 7 carbons

Cells use to create ATP or cellular energy

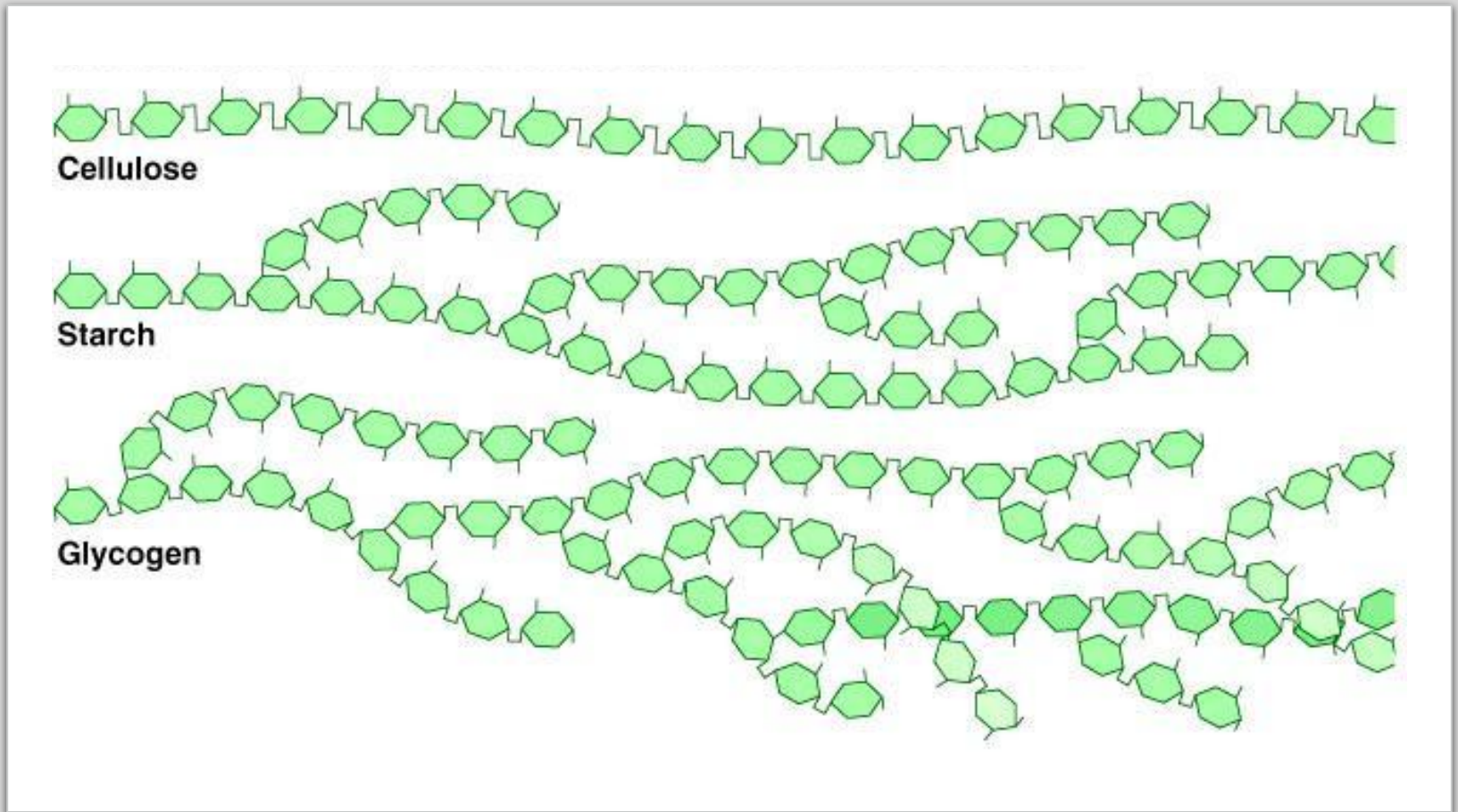
C, H, O ratio is 1:2:1

Glucose, Fructose, Ribose

## Types of carbohydrates



# Polysaccharides



Complex carbohydrates made up of hundreds of monomers linked by *dehydration synthesis*

Examples of  
polysaccharides:

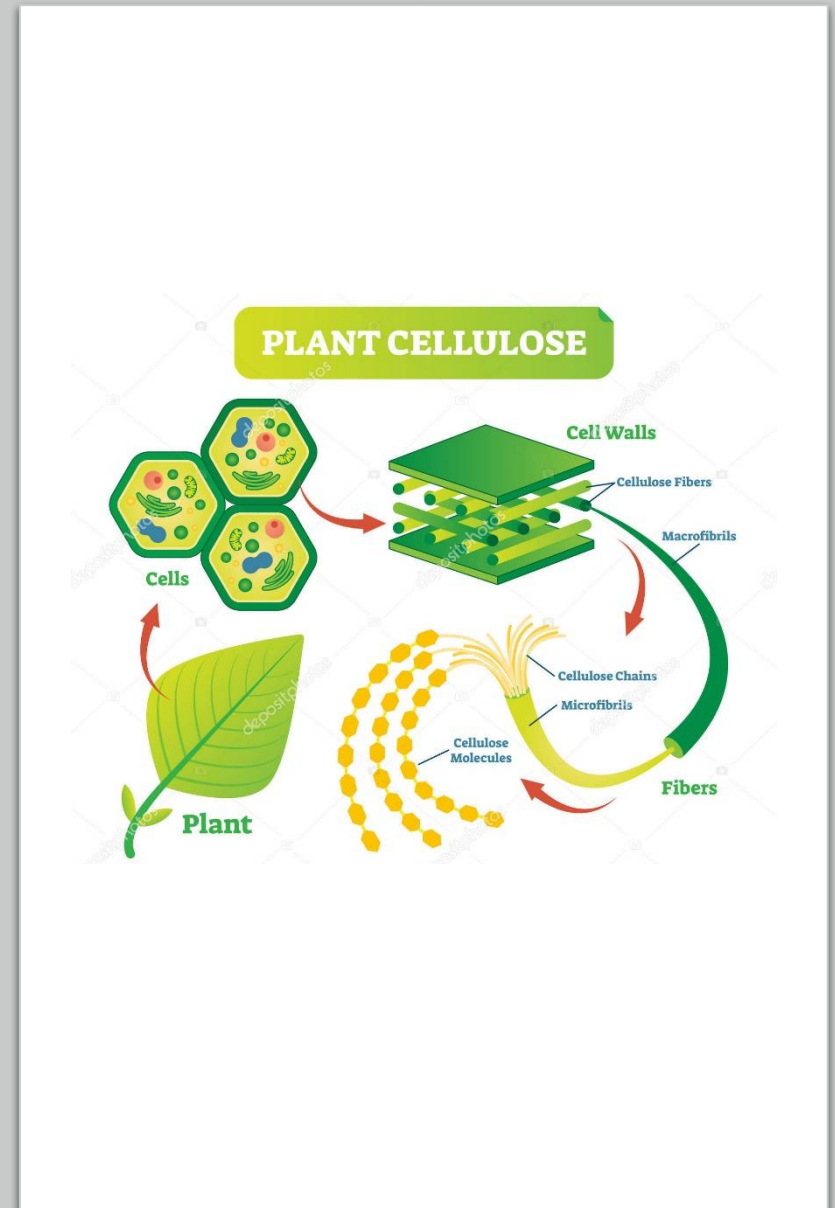
Common polysaccharides include:

- Cellulose - forms wood & parts of plant cell walls.
- Starch - energy storage form in plants
- Glycogen - short term energy storage form in animals.
  
- Cellulose, starch & glycogen are long chains of glucose units; differ in branching patterns

# Examples of polysaccharides

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- Common polysaccharides include:
- Cellulose - forms wood & parts of plant cell walls.

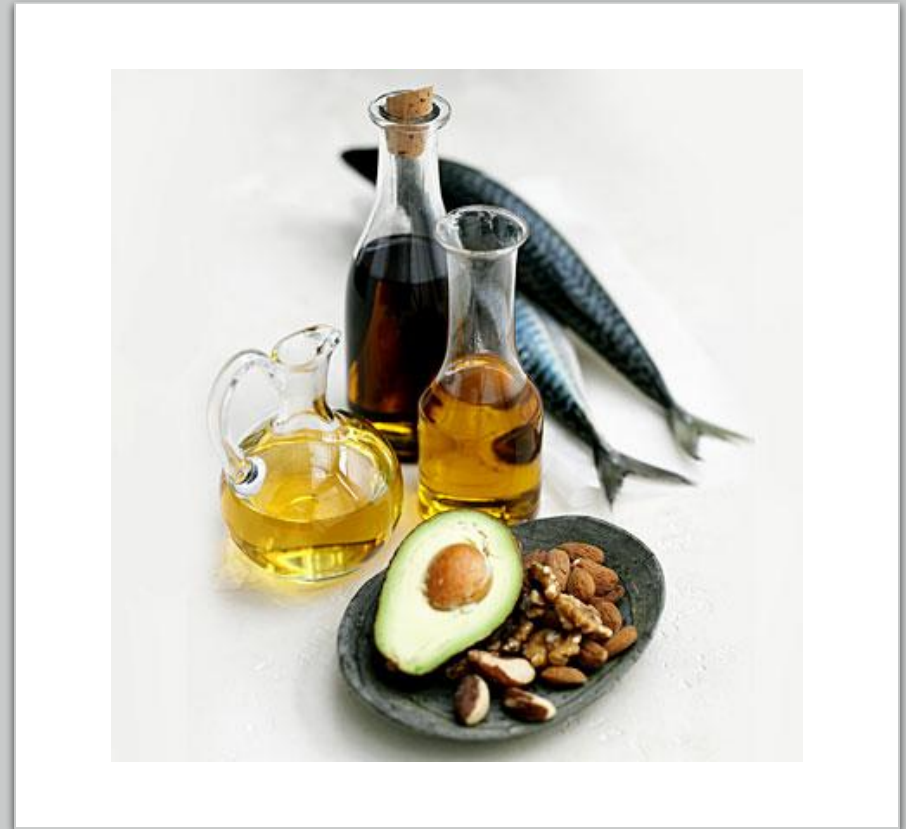


## 2. Lipids

- fatty, waxy, or oily substances in nature
- contain C, H, O
- do not dissolve in water
- Some of the most important are
  - Triglycerides
  - Phospholipids
  - Steroids
  - Waxes

# Lipids in food

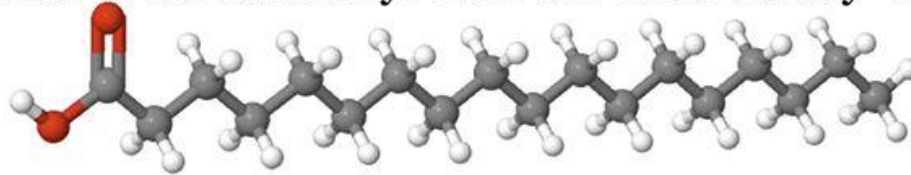
Saturated fats are found in animal products such as butter, cheese, whole milk, ice cream, fatty meats, coconut oil, palm kernel oil.



# Saturated vs. Unsaturated Fatty Acids

All of the chemical bonds in the tails of *saturated fats* are single bonds. This gives the acid the ability to rotate freely and pack together tightly. This is not healthy! The fat can solidify easily.

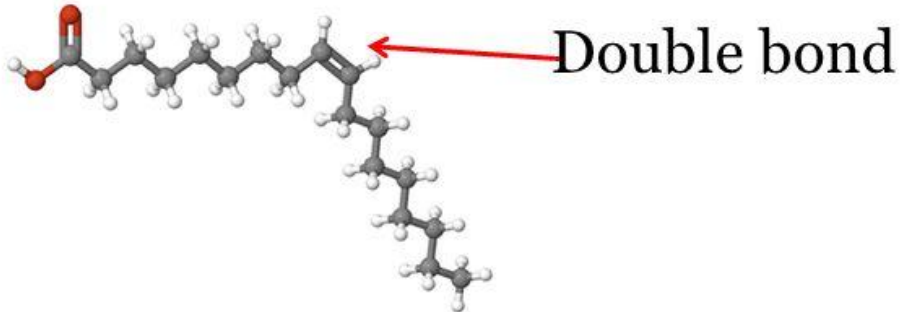
Stearic acid  
(saturated)



(Black atoms = C, white atoms = H, red atoms = O)

*Unsaturated fats* have at least one double bond in their tail. This double bond can be *cis* or *trans* in its orientation.

Oleic acid  
(cis-unsaturated)



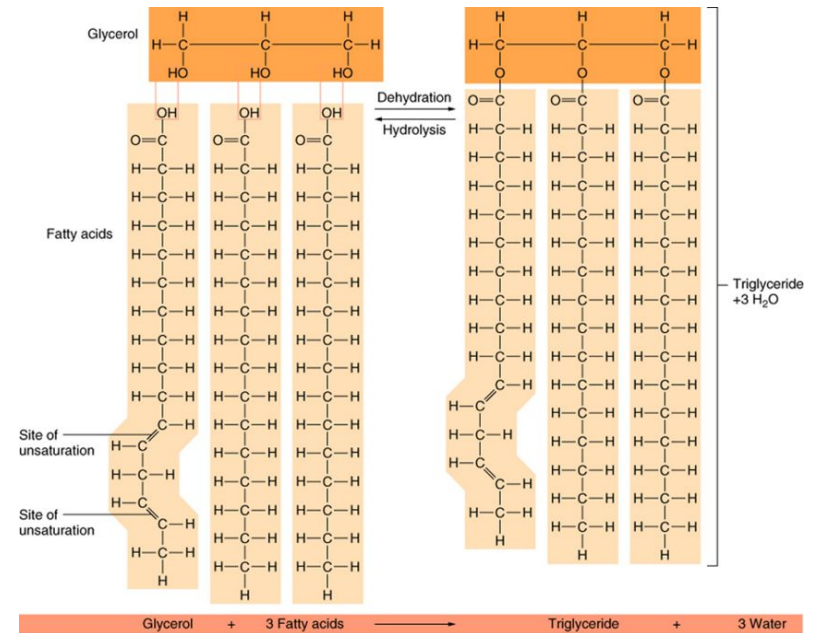




# Example of lipids:

## TRIGLYCERIDES

- composed of glycerol linked to 3 fatty acid chains by *dehydration synthesis*
- They function like cushion organs, insulation and in long-term energy storage as adipose tissue



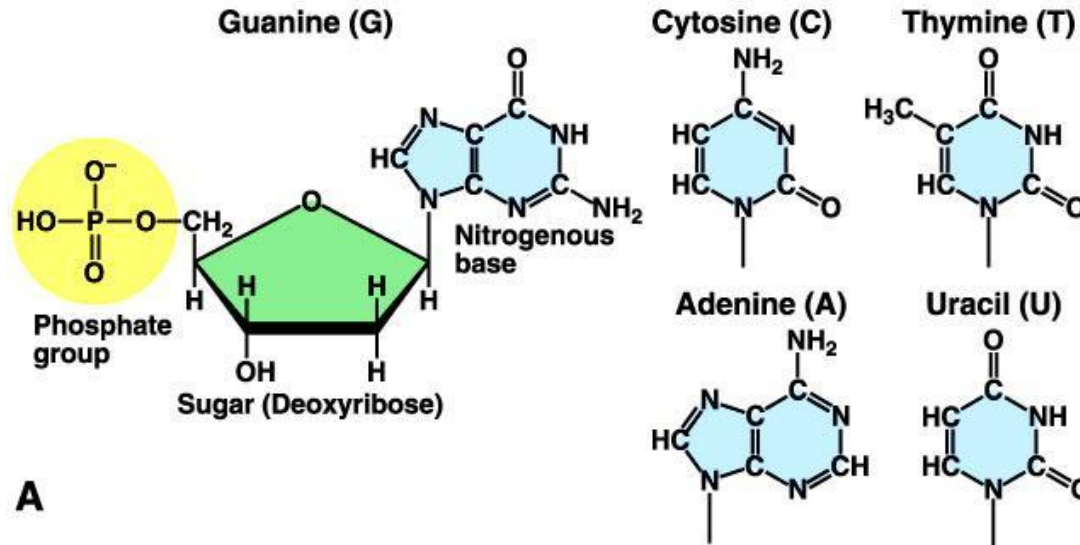
# 4. Nucleic Acids

Nucleic acids contain C, H, O, N, P. Their monomers are nucleotides.

Each nucleotide is composed of:

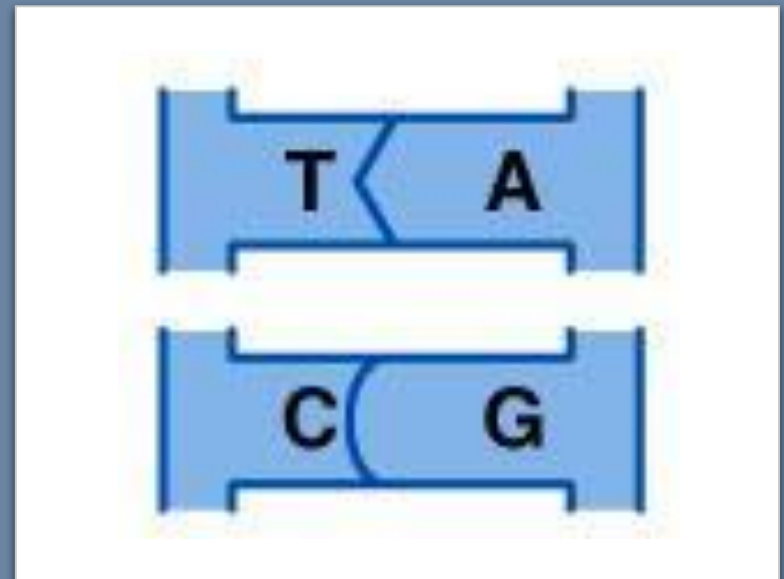
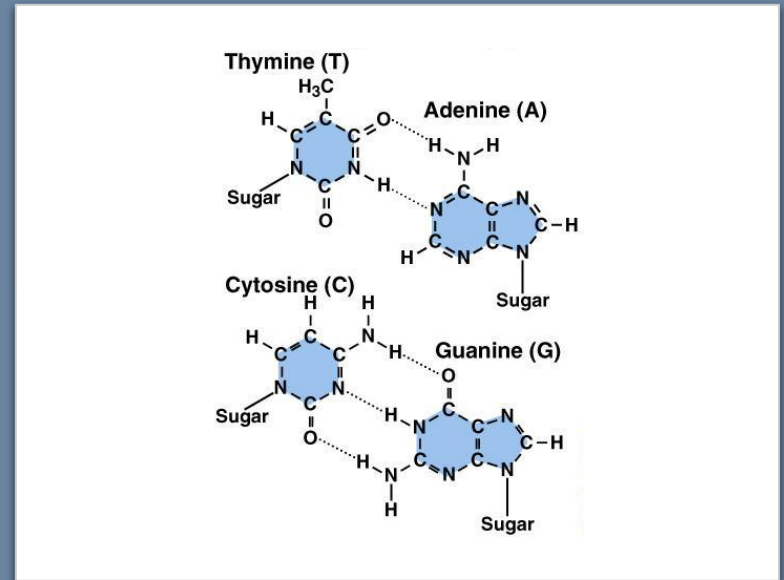
- a 5 carbon sugar (ribose or deoxyribose)
- a phosphate group
- a nitrogenous base (guanine, cytosine, thymine, adenine or uracil).
- They are **DNA, deoxyribonucleic acid** and **RNA, ribonucleic acid**

**Nucleotides**—consist of a sugar (ribose or deoxyribose), a phosphate, and one of five nitrogenous bases.

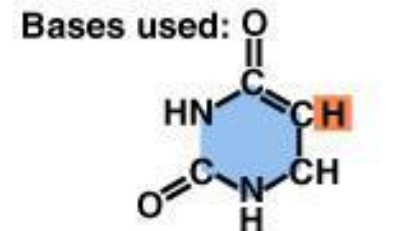
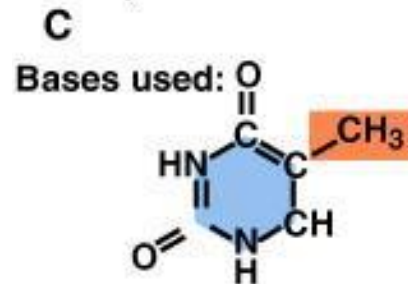
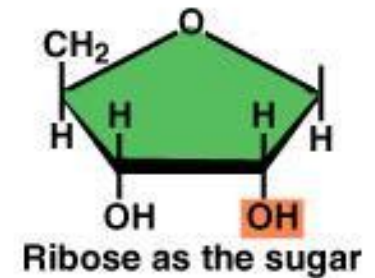
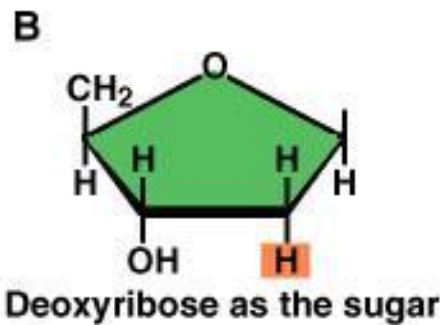
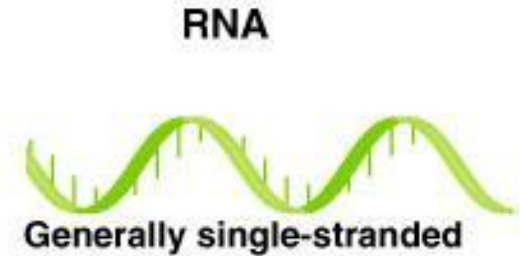
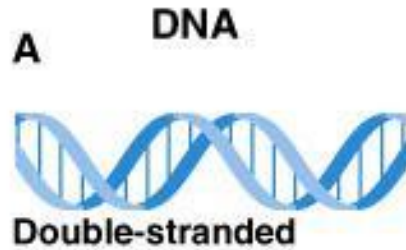


# Pyrimidines (T & C) form hydrogen bonds with purines (A & G)

- **Thymine** pairs with **Adenine**, forming 2 hydrogen bonds
- **Cytosine** pairs with **Guanine**, forming 3 hydrogen bonds



# Comparison of DNA and RNA



Cytosine (C)  
Adenine (A)  
Guanine (G)

Cytosine (C)  
Adenine (A)  
Guanine (G)